## Crystal Structure

# Tris(2,6-di-tert-butyl-4-methylphenol-ato- $O$ )(triphenylphosphine oxide- $O$ )samarium(III) 

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In the title complex, $\left[\mathrm{Sm}\left(\mathrm{C}_{15} \mathrm{H}_{23} \mathrm{O}\right)_{3}\left(\mathrm{C}_{18} \mathrm{H}_{15} \mathrm{OP}\right)\right]$, the central metal atom is coordinated by three O atoms of the aryloxo ligands and by one O atom of the triphenylphosphine oxide molecule resulting in a distorted tetrahedral geometry. The $\mathrm{Sm}-\mathrm{O}_{\mathrm{Ar}}$ distances ( Ar is aryl) range from 2.122 (2) to 2.190 (2) $\AA$ and the $\mathrm{Sm}-\mathrm{O}_{\mathrm{PPh}_{3}}$ bond length is 2.306 (2) $\AA$.

## Comment

In recent years, lanthanide aryloxides have found a variety of applications as homogeneous catalysts for organic reactions (Shibasaki et al., 1997), and as precursors for organolanthanide syntheses (Clark et al., 1994) and high purity oxide materials (Hubert-Pfalzgraf, 1995). The syntheses, structures and reactivities of aryloxo-lanthanide complexes have attracted a great deal of attention and the investigation of the chemistry of lanthanide complexes containing aryloxo ligands has produced complexes exhibiting a variety of structural types (Evans et al., 1997; Deacon et al., 1997). In our study of the syntheses and reactivities of lanthanide aryloxides (Qi et al., 1995; Yao et al., 1998), the title complex, (I), was obtained from the reaction of $\left[(\mathrm{ArO})_{2} \mathrm{Sm}(\mathrm{thf})_{3}\right]\left(\mathrm{ArO}=\mathrm{OC}_{6} \mathrm{H}_{2}\right.$-2,6-di-tert-butyl-4-Me and thf is tetrahydrofuran) with excess triphenylphosphine oxide in toluene.


The crystal structure determination indicates that (I) consists of one Sm atom, three aryloxo ligands and one triphenylphosphine oxide molecule. The Sm atom is coordinated
by four O atoms, resulting in a distorted tetrahedral geometry. The coordination geometry around the central metal atom is similar to that in $\left[\mathrm{Sm}(\mathrm{OAr})_{3}(\mathrm{thf})\right]$ •thf (Qi et al., 1995), but different from that in $\left[\mathrm{Nd}(\mathrm{Odpp})_{3}(\mathrm{thf})\right]$, which has a distorted trigonal-bipyramidal stereochemistry as a result of intramolecular $\pi$-bonding (Deacon et al., 1995).

The $\Sigma \mathrm{O}-\mathrm{Sm}-\mathrm{O}$ of the three aryloxo O atoms around samarium is $345.0^{\circ}$; this value is comparable with those in $\left[\mathrm{Sm}(\mathrm{OAr})_{3}(\mathrm{thf})\right] \cdot \mathrm{thf} \quad\left(351.9^{\circ} ; \quad \mathrm{Qi}\right.$ et al., 1995) and $\left[\mathrm{Nd}(\mathrm{Odpp})_{3}(\mathrm{thf})\right]\left(353.2^{\circ}\right)$. However, the three $\mathrm{O}_{\mathrm{Ar}}-\mathrm{Sm}-$ $\mathrm{O}_{\mathrm{Ar}}$ bond angles range from 99.42 (8) to $138.29(8)^{\circ}$; this is different from the case in $\left[\operatorname{Sm}(\mathrm{OAr})_{3}(\mathrm{thf})\right] \cdot$ thf, where the three bond angles are nearly equal, and this may be attributed to the presence of the triphenylphosphine oxide ligand which is more bulky than the tetrahydrofuran ligand. The three $\mathrm{Sm}-$ $\mathrm{O}_{\mathrm{Ar}}-\mathrm{C}$ bond angles range from 172.9 (2) to 175.3 (2) ${ }^{\circ}$; they are all somewhat larger than those in $\left[\mathrm{Sm}(\mathrm{OAr})_{3}(\mathrm{thf})\right] \cdot$ thf $(\mathrm{Qi}$ et al., 1995) and are comparable with those in $\left[\mathrm{Sm}(\mathrm{OAr})_{2}(\mathrm{thf})_{3}\right] \cdot$ thf $\left(\mathrm{Qi}\right.$ et al., 1994) and $\left[\mathrm{La}\left(\mathrm{OC}_{6} \mathrm{H}_{3} \mathrm{Me}_{2^{-}}\right.\right.$ $\left.2,6)_{3}\left\{\mathrm{MeO}\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{O}\right)_{4} \mathrm{Me}\right\}\right]$ (Aspinall \& Williams, 1996). The $\mathrm{Sm}-\mathrm{O}-\mathrm{P}$ angle of $170.06(14)^{\circ}$ is comparable with those found for lanthanide-triphenylphosphine oxide complexes (Evans et al., 1985; Deacon et al., 1989).

The three $\mathrm{Sm}-\mathrm{O}_{\mathrm{Ar}}$ bond lengths range from 2.112 (2) to 2.190 (2) $\AA$, with an average of $2.155 \AA$; this value is similar to that of $\left[\mathrm{Sm}(\mathrm{OAr})_{3}(\mathrm{thf})\right] \cdot$ thf $(2.151 \AA$; Qi et al., 1995). Subtraction of the estimated radius for four-coordinate trivalent samarium ( $0.84 \AA$ A Shannon, 1976) from the $\mathrm{Sm}-\mathrm{O}$ distance gives a value of $1.31 \AA$, which is in the range 1.24 $1.31 \AA$ found for a similar treatment of $\mathrm{Ln}-\mathrm{O}_{\mathrm{Ar}}$ distances in


Figure 1
The molecule of (I) showing $40 \%$ probability displacement ellipsoids. H atoms have been omitted for clarity.
other aryloxo-lanthanide complexes (Hitchcock et al., 1983, 1987). The $\mathrm{Sm}-\mathrm{O}_{\mathrm{PPh}_{3}}$ distance of 2.306 (2) $\AA$ is comparable with the values in $\left[\left(\mathrm{C}_{5} \mathrm{Me}_{5}\right)_{2}\left\{\left(\mathrm{C}_{6} \mathrm{H}_{5}\right)_{3} \mathrm{PO}\right\} \mathrm{Sm}\right]_{2}(\mu-$ $\mathrm{OCH}=\mathrm{CHO})\left(\right.$ Evans et al., 1985) and $\left[\left(\mathrm{C}_{5} \mathrm{H}_{5}\right)_{2} \mathrm{Yb}\left(\mathrm{OPPh}_{3}\right)_{2}\right]$ (Deacon et al., 1989) if the difference in the ionic radii is considered.

## Experimental

The title complex is very sensitive to air and moisture. Consequently, all manipulations were carried out under argon by Schlenk techniques. Crystals were obtained from the reaction of $\left[\mathrm{Sm}(\mathrm{OAr})_{2}(\mathrm{thf})_{3}\right]$ with excess triphenylphosphine oxide in toluene at 353 K , followed by crystallization from a toluene solution at 263 K . A crystal suitable for single-crystal structure determination was sealed in a thin-walled glass capillary.

## Crystal data

$\left[\mathrm{Sm}\left(\mathrm{C}_{15} \mathrm{H}_{23} \mathrm{O}\right)_{3}\left(\mathrm{C}_{18} \mathrm{H}_{15} \mathrm{OP}\right)\right]$
$M_{r}=1086.62$
Monoclinic, $P 2_{1} / n$
$a=11.682(2) \AA$
$b=38.168$ (6) $\AA$
$c=14.245(2) \AA$
$\beta=112.630(10)^{\circ}$
$V=5862.5(16) \AA^{3}$
$Z=4$

## Data collection

Siemens $P 4$ diffractometer
$\omega$ scans
Absorption correction: $\psi$ scan (North et al., 1968) $T_{\text {min }}=0.851, T_{\text {max }}=0.984$
11426 measured reflections
10332 independent reflections
6845 reflections with $I>2 \sigma(I)$

$$
\begin{aligned}
& D_{x}=1.231 \mathrm{Mg} \mathrm{~m}^{-3} \\
& \text { Mo } K \alpha \text { radiation } \\
& \text { Cell parameters from } 38 \\
& \quad \text { reflections } \\
& \theta=2.99-16.50^{\circ} \\
& \mu=1.072 \mathrm{~mm}^{-1} \\
& T=297(2) \mathrm{K} \\
& \text { Prismatic, yellow } \\
& 0.50 \times 0.40 \times 0.24 \mathrm{~mm} \\
& \\
& R_{\text {int }}=0.025 \\
& \theta_{\max }=25^{\circ} \\
& h=0 \rightarrow 13 \\
& k=0 \rightarrow 45 \\
& l=-16 \rightarrow 15 \\
& 3 \text { standard reflections } \\
& \quad \text { every } 97 \text { reflections } \\
& \text { intensity decay: } 2.94 \%
\end{aligned}
$$

## Refinement

Refinement on $F^{2}$

$$
R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.033
$$

$$
w R\left(F^{2}\right)=0.057
$$

$$
S=0.972
$$

$$
10320 \text { reflections }
$$

$$
644 \text { parameters }
$$

$$
\begin{aligned}
& \text { 644 parameters } \\
& \text { H-atom parameters constrained }
\end{aligned}
$$

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{o}^{2}\right)+(0.0246 P)^{2}\right] \\
& \quad \text { where } P=\left(F_{o}^{2}+2 F_{c}^{2}\right) / 3 \\
& (\Delta / \sigma)_{\max }=-0.006 \\
& \Delta \rho_{\max }=0.31 \mathrm{e} \AA^{-3} \\
& \Delta \rho_{\min }=-0.31 \mathrm{e} \AA^{-3} \\
& \text { Extinction correction: SHELXL93 } \\
& \text { Extinction coefficient: } \\
& \quad 2.7(3) \times 10^{-4}
\end{aligned}
$$

All H atoms were generated geometrically $(\mathrm{C}-\mathrm{H}=0.96 \AA)$ and not refined.

Data collection: XSCANS (Siemens, 1994); cell refinement: XSCANS; data reduction: SHELXTL (Siemens, 1994); program(s) used to solve structure: SHELXS86 (Sheldrick, 1990); program(s)

Table 1
Selected geometric parameters $\left(\AA^{\circ},{ }^{\circ}\right)$.

| $\mathrm{Sm}-\mathrm{O} 4$ | $2.112(2)$ | $\mathrm{Sm}-\mathrm{O} 1$ | $2.306(2)$ |
| :--- | ---: | :--- | :---: |
| $\mathrm{Sm}-\mathrm{O} 2$ | $2.163(2)$ | $\mathrm{P}-\mathrm{O} 1$ | $1.509(2)$ |
| $\mathrm{Sm}-\mathrm{O} 3$ | $2.190(2)$ |  |  |
| $\mathrm{O} 4-\mathrm{Sm}-\mathrm{O} 2$ | $99.42(8)$ | $\mathrm{O} 3-\mathrm{Sm}-\mathrm{O} 1$ | $94.86(8)$ |
| $\mathrm{O} 4-\mathrm{Sm}-\mathrm{O} 3$ | $107.30(8)$ | $\mathrm{P}-\mathrm{O} 1-\mathrm{Sm}$ | $170.06(14)$ |
| $\mathrm{O} 2-\mathrm{Sm}-\mathrm{O} 3$ | $138.29(8)$ | $\mathrm{C} 19-\mathrm{O} 2-\mathrm{Sm}$ | $174.3(2)$ |
| $\mathrm{O} 4-\mathrm{Sm}-\mathrm{O} 1$ | $127.73(8)$ | $\mathrm{C} 34-\mathrm{O} 3-\mathrm{Sm}$ | $172.9(2)$ |
| $\mathrm{O} 2-\mathrm{Sm}-\mathrm{O} 1$ | $93.36(8)$ | $\mathrm{C} 49-\mathrm{O} 4-\mathrm{Sm}$ | $175.3(2)$ |

used to refine structure: SHELXL93 (Sheldrick, 1993); molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: SK1411). Services for accessing these data are described at the back of the journal.

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